BASIC CLASS NOTES

NOVEMBER 18, 2015

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Reading Review and Class Preparation This should be filled out prior to class.	
Key Concepts to Be Discussed in Class:	
Questions About Subject Matter for Class Session:	
 So What? Why? Who Cares? Quite often Making Something Requires Joining the Pieces Togethe Techniques other Than Mechanical Fastening are Often Used 	er

- A Problem Correctly Stated is a Problem Half Solved \circ Charles Kettering

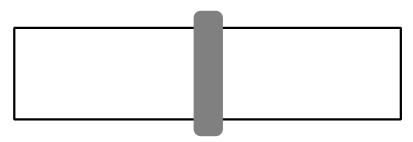
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BASIC CLASS NOTES

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<u>Definition of Brazing</u>

Joint forms Without Melting Base Materials



Ref: Brazing Handbook; AWS (1991)

• Filler Metal Must Wet the Base Metal Surfaces and Be Drawn into Joint by Capillary Action

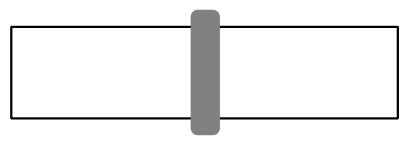
• The Lowest Temperature that the Filler Metal is Completely Liquid Must Be at Least 450°C

BASIC CLASS NOTES

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<u>Definition of Soldering</u>

• Joint forms Without Melting Base Materials



Ref: Brazing Handbook; AWS (1991)

• Filler Metal Must Wet the Base Metal Surfaces and Be Drawn into Joint by Capillary Action

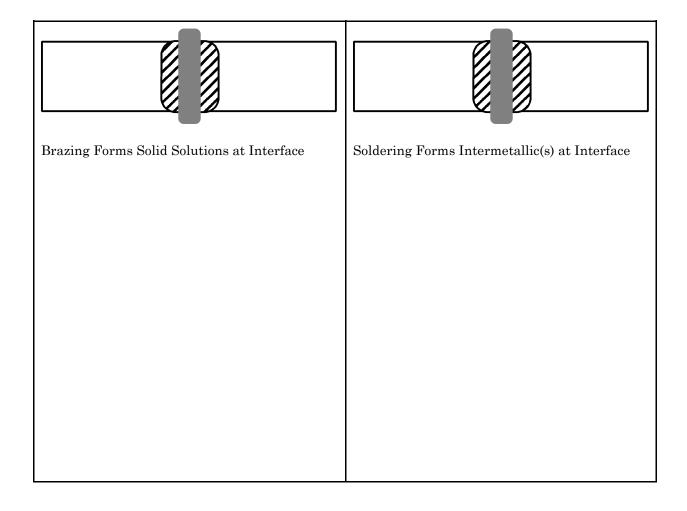
 \bullet The Lowest Temperature that the Filler Metal is Completely Liquid Must Be Less Than $450^{\circ}\mathrm{C}$

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Temperature 450°C

- Historical
 - 0 Based on Joining of Steel
 - Brazes Based on Cu-Zn Alloys of Brasses 0
 - 0 Solders Based on Sn Based Alloys
- Practical
 - Lowest T Braze Al-4Cu-10Si ($T_{\rm M} = 524^{\circ}{\rm C}$) Highest T Solder Au-3Si ($T_{\rm M} = 363^{\circ}{\rm C}$)
- Scientific

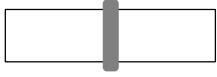


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Fluxes in Brazing

 Filler Metal Must Wet the Base Metal Surfaces and Be Drawn into Joint by Capillary Action



Ref: <u>Brazing Handbook</u>:AWS (1991)

• Oxide Formation Must Be Prevented

• You Will Need to Add Flux When Brazing

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Brazing or Welding?

- Two Pieces of 1040 Steel
 - Joined With Copper Filler Metal
- 1040 Steel has a Higher Melting Temperature than Copper
- Copper has a Higher Melting Temperature than 70-30 Brass

- Two Pieces of Copper
 - Joined With Copper Filler Metal

- Two Pieces of Copper
 - Joined With 70-30 Brass Filler Metal

- A Piece of Copper and a Piece of 1040 Steel
 - Joined With Copper Filler Metal

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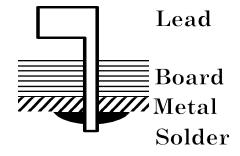
Sample Solder Alloys

- Mostly Tin (Sn) Based
- Historical Workhorse Lead(Pb)-Tin(Sn)

Solder	Tm (C)	Shear Strength (MPa)	Ductility %
Solder	Tm (C)	(IVIPa)	70
Sn-Pb	183	28	1.3
Sn-Ag	221	30	0.69
Sn-Bi	138	24	1.3
95-5	280	30	18.3

Importance of Solder Joints

- Provides Electrical Contact
 - o Requires Metallurgical Bond
 - o Strength / Fatigue Resistance Required
- Critical for Electronic Performance
 - Unglorified Part of Semiconductor Revolution
 - "For Want of a Nail ... A Kingdom was Lost"

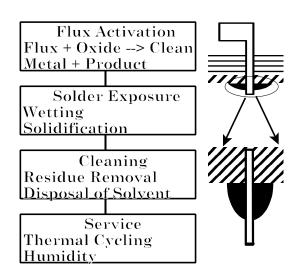


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Production of Solder Joints

- Form Contact Between Lead and Metal
 - Remove any Oxide or Impurity
 - o Cause Molten Solder to Wet Metal
 - o Solidify Solder
 - o Remove any Residue



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<u>Miniaturization of Joints</u>

- Increasing Amount of SMT
- Paste
 - $\circ \qquad \quad \text{Combination of Alloy Powder and Flux}$
 - o Identified Based On Through-Hole Experience



Through-Hole Assembly



Surface Mount Assembly

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<u>Low Temperature Problems</u>

- Cleaning Required

 Prior to Soldering

More Aggressive Fluxes are Needed

- Very High Thermal Stress
 - Solder is in Tension

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Summary

- Welding
 - \circ Base Metal and Filler Metal Melt
 - Strength Considerations
 - Fuel Sources
 - o Oxide Prevention

- Brazing / Soldering
 - Only Filler Metal Melts
 - o Same Considerations as In Welding
 - o Problem With Metal Combinations

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Polymer Joining and Adhesive Bonding